Exam 3

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200 points total

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25 pts 1):

The table below lists several simple chemical tests in the rows and some organic compounds in the columns. Place a "+" in the table to indicate a positive result for the test and a "0" to indicate a negative test. (Consider solubility a positive test and insolubility a negative test for H_2O , H_2SO_4 , HCl, and NaOH. A positive test for $NaHCO_3$ is evolution of CO_2 bubbles).

	ethanol	В	С	D	E
H ₂ O solubility	+	0	0	0	0
aq HCl solubility	+			+	(
aq NaOH solubility	+	1	0	. 0	+
conc H ₂ SO ₄ solubility	+	+	+	+	+
NaHCO ₃	\bigcirc	+	0	\mathcal{O}	0
Decolorizes Br ₂ / CCl ₄	0	+	+	+	0

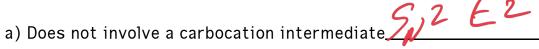
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30pts 2) For each statement below, write S_N1 , S_N2 , E1, or E2 if the statement applies to that mechanism.(Some statements are applicable to more than one mechanism)



b) Is first order in haloalkane and zero order in nucleophile

c) Involves inversion of configuration at site of substitution $\frac{5}{2}$

d) Is first order in haloalkane and first order in base $\frac{1}{2}$

e) Rearrangements may occur

f) Order of reactivity of haloalkanes is 3° > 2° > 1°

25 pts 3) Circle the compound in each pair that you would expect to react more rapidly as a substrate via S_N2 mechanism:

a) CH₃CH₂CH₂Br or (CH₃)₂CHBr

b) CH₃CH₂CH₂CH₂F or CH₃CH₂CH₂Br

c) (CH₃)₂CHNH₂ or CH₃CH₂CH₂Cl

d) $(CH_3CH_2)_2CHOH$ or $(CH_3CH_2)_2CHOCOCH_3$

e) CH₃CH₂I or CH₃CH₂OH

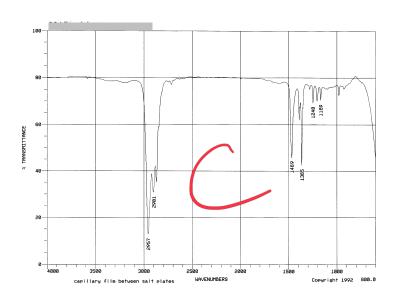
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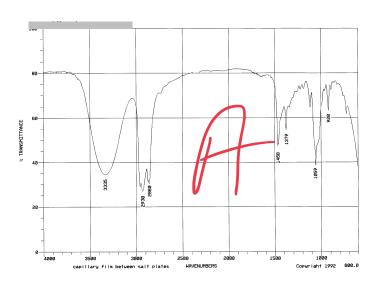
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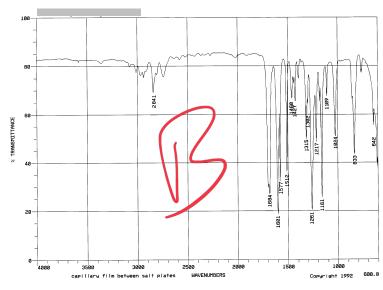
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30 pts 4) For the structures below, associate the structure with its IR spectrum (place the correct letter prominently on the spectrum).







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20 pts 5) Show the expected major product for the following S_N2 reactions:

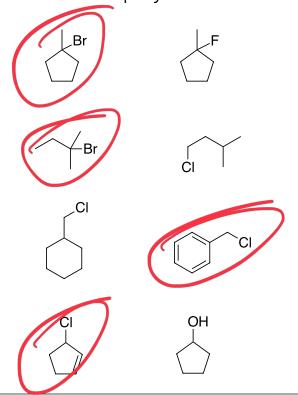
$$CI + NH_{2}^{1-} \longrightarrow NH_{2}$$

$$+ CH_{3}CO_{2}CH_{3} \longrightarrow O - CH_{3}$$

$$CI + OH \longrightarrow MH_{2}$$

$$+ CH_{3}CO_{2}CH_{3} \longrightarrow O - CH_{3}$$

20 pts 6) Circle the compound in each pair that would undergo solvolysis in aqueous methanol more rapidly:



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20 pts 7) a) Draw the structure of 2,3,5,5-tetramethyl-2-hexene:

XX

b) Give the IUPAC names for the compound on the left and state whether the compound on the right is (Z)- or (E)-

G-ch 600-44 dimethyl2-he xene L CH3

30 pts 8) Show the expected major products of the reactions below:

$$\begin{array}{c}
Br \\
CH_3OH
\end{array}$$

$$\begin{array}{c}
OH \\
H_2SO_4 \\
heat
\end{array}$$

$$\begin{array}{c}
H_2O
\end{array}$$

$$\begin{array}{c}
H_2O
\end{array}$$

$$\begin{array}{c}
OH
\end{array}$$

$$\begin{array}{c}
OH
\end{array}$$

$$\begin{array}{c}
H_2O
\end{array}$$

$$\begin{array}{c}
OH
\end{array}$$

$$\begin{array}{c}
OH$$

$$\begin{array}{c}
OH
\end{array}$$

$$\begin{array}{c}
OH
\end{array}$$

$$\begin{array}{c}
OH
\end{array}$$

$$\begin{array}{c}
OH$$

$$\begin{array}{c}
OH
\end{array}$$

$$\begin{array}{c}
OH
\end{array}$$

$$\begin{array}{c}
OH$$

$$\begin{array}{c}
OH$$

$$OH$$

$$\begin{array}{c}
OH$$

$$OH$$

$$OH$$